

Modelling the atmospheric chlorine 36 dry deposition on grass using micrometeorology

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Chlorine radioactive isotope

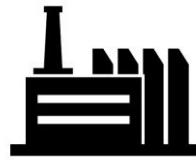
Long-lived radionuclide ($t = 301,000$ years)

$$\text{Activity} = 1,22 \times 10^9 \text{ Bq.g}^{-1}$$



*High solubility
and
bioavailability*

(Bastviken et al., 2013)



*Natural
and
anthropogenic
sources*

(Pupier et al., 2016)



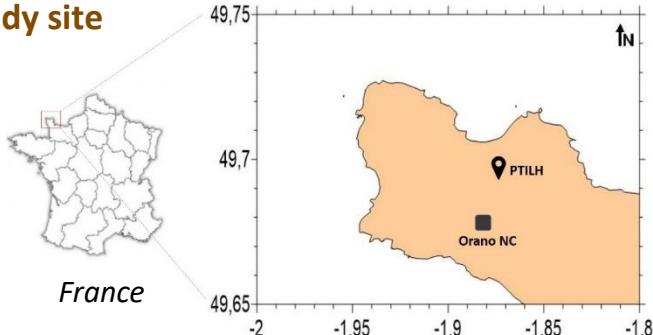
*Lack of data on his
fate and
dosimetric impact*

(Tanaka and Thiry, 2019)
(Le Dizès and Gonze, 2019)

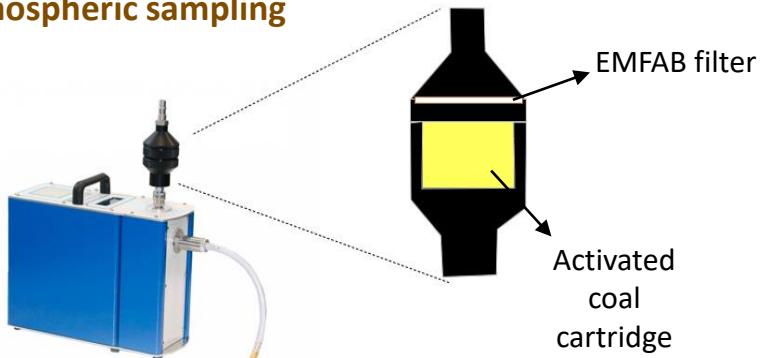
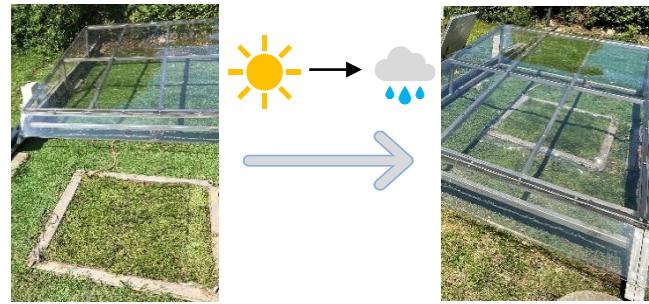


*Wet and dry
deposition*

(Lazarev, 2003)

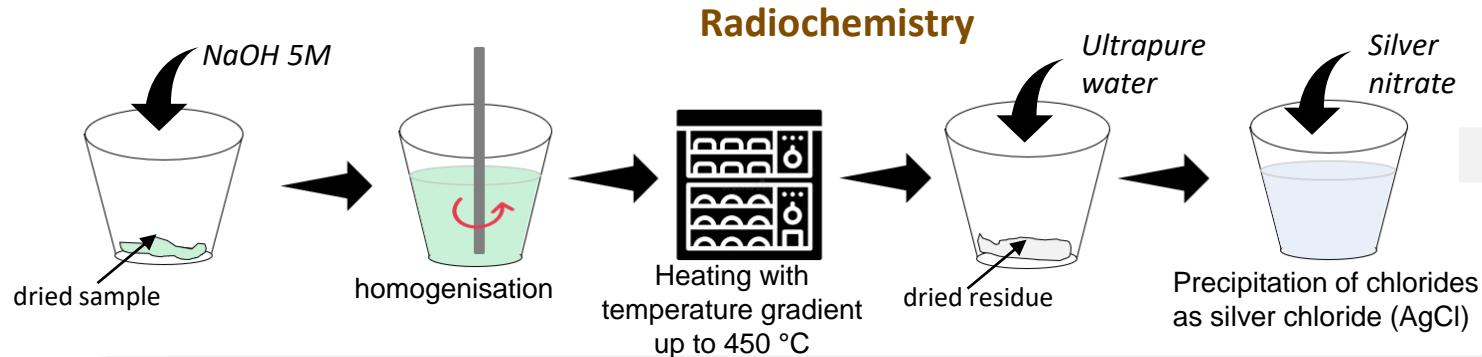
Study site

*PTILH : IRSN La Hague technical platform
Orano NC : Nuclear fuels reprocessing plant*

Atmospheric sampling**Monitoring of weather parameters****Grass sampling**

Sunny weather

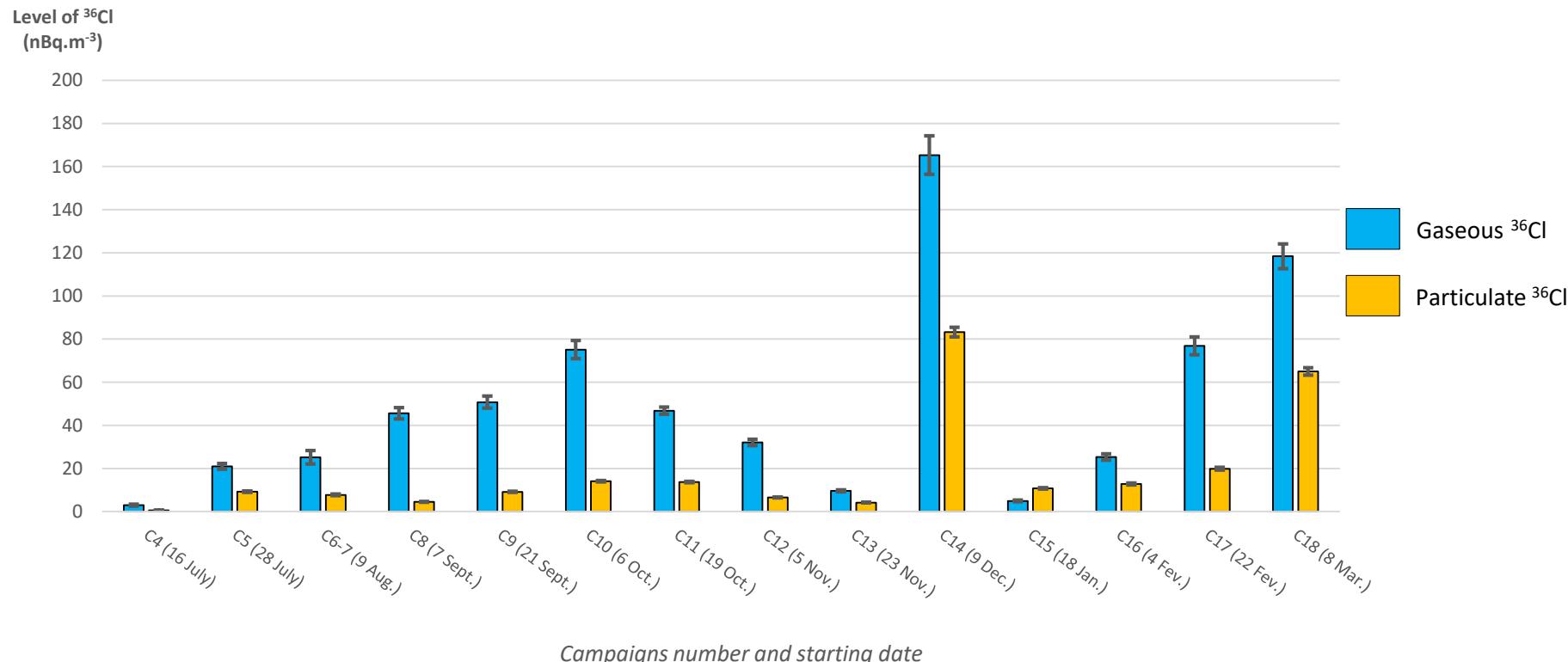
Rainy weather



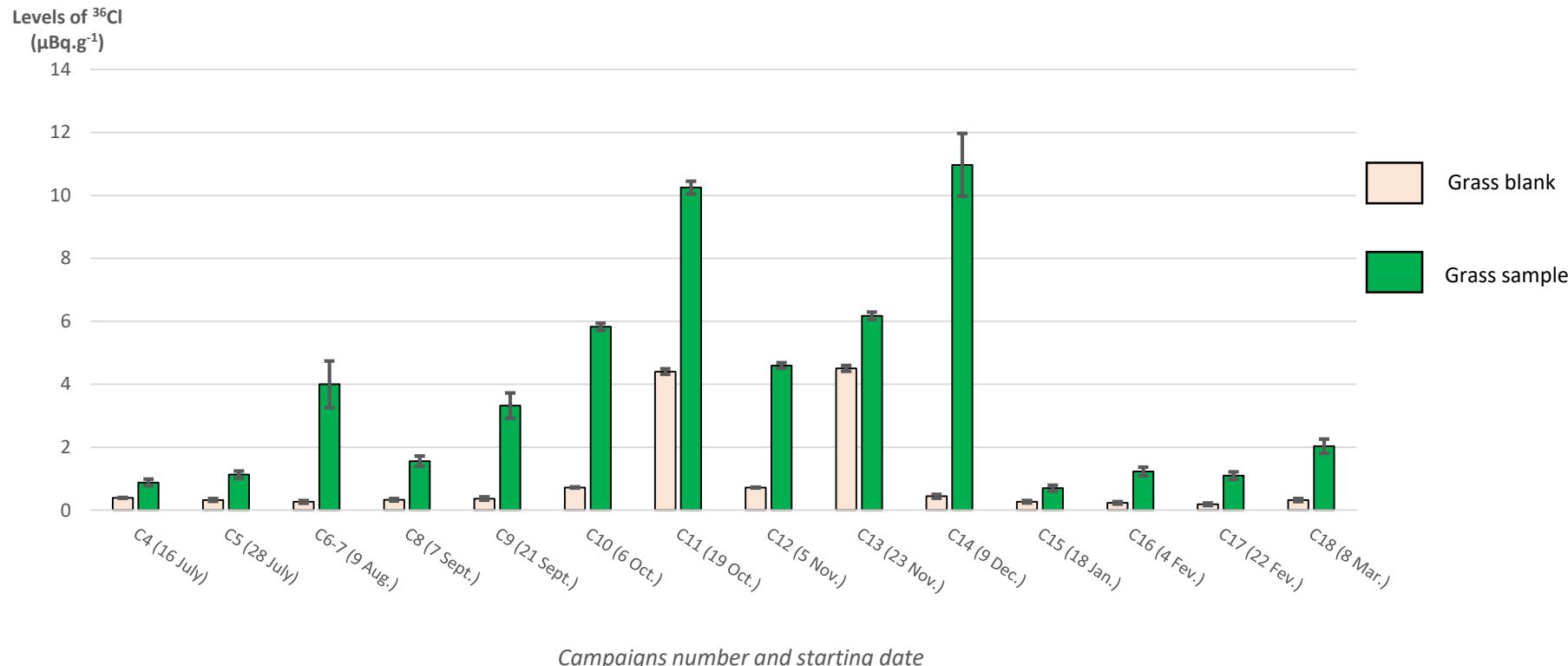
Accelerated Mass spectrometer (**AMS**) - Accelerator for Environmental, Earth and Risk Sciences (**ASTER**)



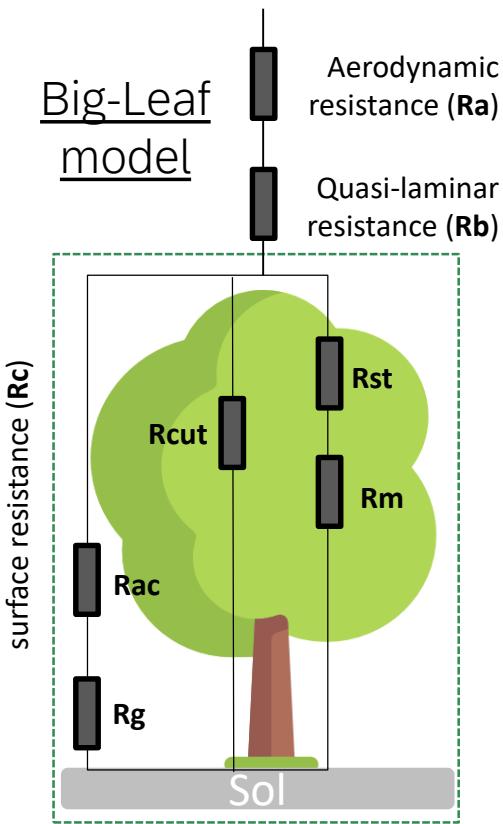
Atmospheric chlorine 36



Chlorine 36 in grass samples



Modelling gaseous ^{36}Cl



(adapted from Zhang et al., 2003)

$$\nu_d = \frac{1}{\sum R}$$

$$\nu_d = \frac{1}{R_a + R_b + R_c}$$

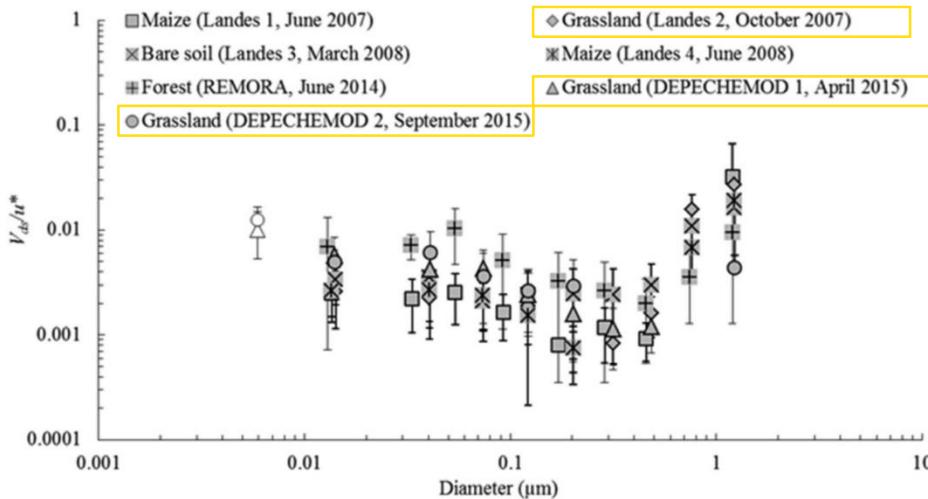
—~\~\~— **Aerodynamic resistance (R_a)** is the resistance to turbulent transport of matter through the atmospheric surface layer. It is common to all gases

—~\~\~— **Quasi-laminar resistance (R_b)** is the resistance to material transfer at the latter layer to the surface. This resistance is due to molecular diffusion in the thin sub-layer near a flat rough surface

—~\~\~— **Canopy or surface resistance (R_c)** is the total resistance to gas transfer at the surface and at the substrate. Surface resistance refers to the resistance at the sheet level

Modelling particulate ^{36}Cl

- Diameter of the particle (d_p)
- Atmospheric stability ($1/L$)
- Air friction velocity (u^*)



(Pellerin et al., 2017)

Particles dry deposition velocity



Low pressure Impactor (LPI) DEKATI



$$D_{\text{particles}} = 0,48 \mu\text{m}$$

Particles diameter

A

B

If stable or neutral atmosphere

$$\frac{v_d}{u^*} = A$$

If unstable atmosphere

$$\frac{v_d}{u^*} = A \left(1 + \left(\frac{B}{L} \right)^{\frac{2}{3}} \right)$$

(Damay et al., 2009)

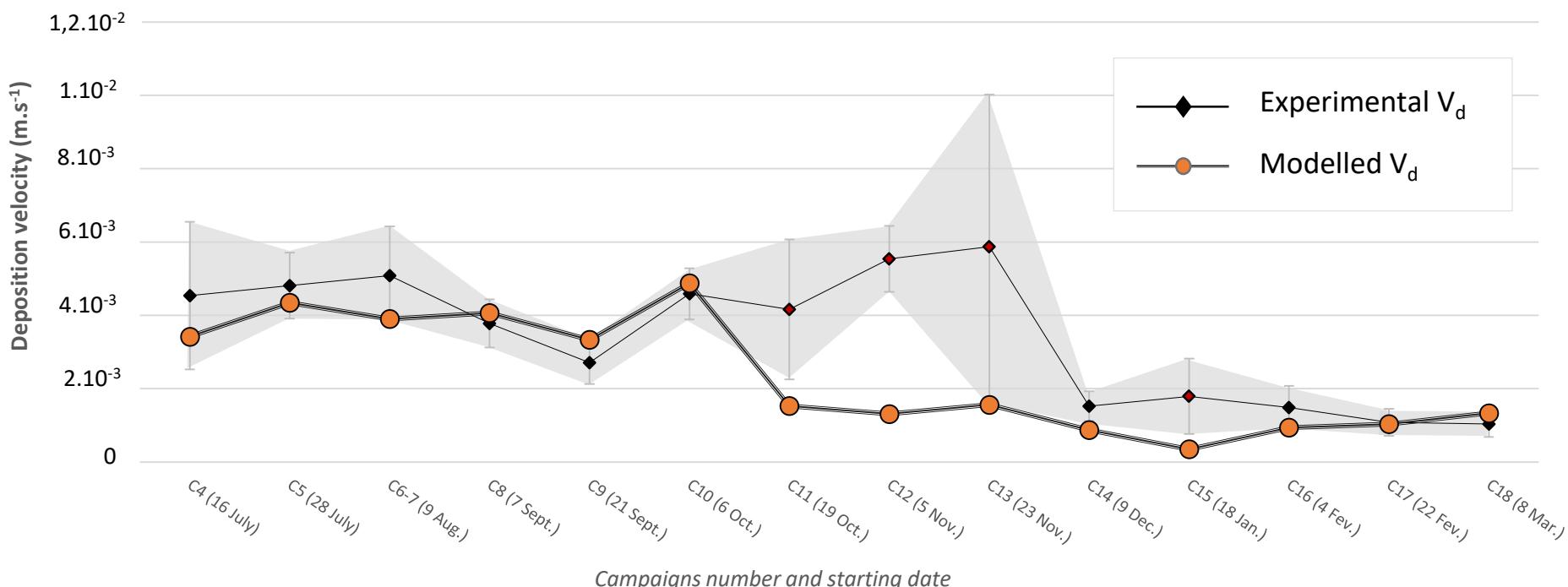
(Pellerin et al., 2017)

Experimental and modelled dry deposition velocities

$$Vd_{\text{gas}} = 3.10^{-4} \sim 5.10^{-3} \text{ m.s}^{-1}$$

$$Vd_{\text{particles}} = 6.10^{-5} \sim 9.10^{-4} \text{ m.s}^{-1}$$

$^{36}\text{Cl}_{\text{gaz}} > 97\%$ of deposits



Take-home messages

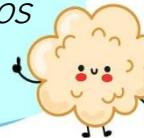
Dry deposition velocities of ^{36}Cl on grass vary with the season and parameters related to atmospheric turbulence

The deposition models: Big-leaf for gaseous ^{36}Cl and Damay-Pellerin for particulate ^{36}Cl developed can provide a good estimate of dry deposition

^{36}Cl deposits are dominated by the gaseous fraction but particulate ^{36}Cl with larger diameter may question this trend

Thanks for your attention!

More infos



Abstract



Project



Contact

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